



DEPARTMENT OF CHEMISTRY

COURSE PLAN – PART I			
Name of the programme and specialization	B.Tech. I Semester- Electrical and Electronics Engineering		
Course Title	Chemistry (Lab)		
Course Code	CHIR 12	No. of Credits	2
Course Code of Pre-requisite subject(s)	Nil		
Session	January 2020	Section (if, applicable)	A
Name of Faculty	Dr. Shima P Damodaran	Department	Chemistry
Official Email	shima@nitt.edu	Telephone No.	9447956884
Name of Course Coordinator(s)(if, applicable)	Dr. Shima P Damodaran		
Official E-mail	shima@nitt.edu	Telephone No.	9447956884
Course Type (please tick appropriately)	<input checked="" type="checkbox"/> Core course	<input type="checkbox"/> Elective course	
<b>Syllabus (approved by BOS)</b>			
<b>LIST OF EXPERIMENTS</b>			
<ol style="list-style-type: none"> <li>1. Estimation of carbonate, non-carbonate and total hardness in the given water sample.</li> <li>2. Estimation of dissolved oxygen in the given water sample.</li> <li>3. Determination of the percentage of Fe in the given steel sample.</li> <li>4. Estimation of Fe<sup>3+</sup> by spectrophotometer.</li> <li>5. Corrosion rate by polarization technique</li> <li>6. Conductometric titration</li> <li>7. Potentiometric titration</li> <li>8. pH-metric titration</li> <li>9. Percentage purity of bleaching powder</li> <li>10. Determination of molecular weight of the polymer by Viscometry</li> <li>11. Study of three component system.</li> <li>12. Demonstration experiments using Advanced Spectroscopic Techniques, (UV-Vis, FTIR, Raman)</li> </ol>			
<b>Reference Books</b>			
<ol style="list-style-type: none"> <li>1. Laboratory Manual, Department of Chemistry, National Institute of Technology, Tiruchirappalli.</li> <li>2. S.K. Bhasin, S. Rani, Laboratory Manual on Engineering Chemistry, Dhanpat Rai Publishing Company, New Delhi, 2011.</li> </ol>			
<b>COURSE OBJECTIVES</b>			
To introduce the student's the experiments on (i) estimation of total hardness and (ii) dissolved oxygen in a given water sample, (iii) determination of the percentage of Fe in the given steel sample, (iv) estimation of Fe <sup>3+</sup> by spectrophotometer, (v) determination of corrosion rate by polarization technique, (vi) conductometric titration, (viii) potentiometric titration, (viii) pH-metric titration, (ix) determination of percentage purity of bleaching powder, (x) determination of molecular weight of the polymer by viscometry			



MAPPING OF COs with POs	
Course Outcomes	Programme Outcomes (PO) (Enter Numbers only)
<ul style="list-style-type: none"> <li>The chemistry laboratory course will consist of experiments illustrating the principles of chemistry relevant to the study of science and engineering.</li> <li>The students will learn how to estimate various components from the corresponding bulk mixture</li> </ul>	1,2,14,9

**COURSE PLAN – PART II**

**COURSE OVERVIEW**

This is a 2 credit course offered to I year B.Tech students. One lab session (3 h) will be conducted per week. Students will perform experiments illustrating the principles of chemistry relevant to the study of science and engineering and will learn how to estimate various components from the corresponding bulk mixture.

**COURSE TEACHING AND LEARNING ACTIVITIES**

S. No.	Week/Contact Hours	Topic	Mode of Delivery
1	III week of January	Demonstration of experiments	Experiment
2	IV week of January	<ol style="list-style-type: none"> <li>1. Estimation of carbonate, non-carbonate and total hardness in the given water sample.</li> <li>2. Estimation of dissolved oxygen in the given water sample.</li> <li>3. Determination of the percentage of Fe in the given steel sample.</li> <li>4. Estimation of Fe<sup>3+</sup> by spectrophotometer.</li> <li>5. Corrosion rate by polarization technique</li> <li>6. Conductometric titration</li> <li>7. Potentiometric titration</li> <li>8. pH-metric titration</li> <li>9. Percentage purity of bleaching powder</li> <li>10. Determination of molecular weight of the polymer by Viscometry</li> </ol>	Experiment
3	I week of February	"	"
4	II week of February	"	"
5	III week of February	"	"
6	IV week of February	"	"
7	I week of March	"	"
8	II week of March	"	"
9	III week of March	"	"



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10	IV week of March	''	''
11	I week of April	''	''
12	II week of April	''	''
13	III week of April	Compensatory lab	
14	IV week of April		''

## COURSE ASSESSMENT METHODS

S.N o.	Mode of Assessment	Week/Date	Duration	% Weightage
1	Assessment 1 (based on individual experiments done during lab session)	III week of January to III week of April	3 h/week	60
2	Final Assessment	IV week of April	3 hours	40

**Total (100)**

## COURSE EXIT SURVEY

1. Feedback from students during class committee meetings.
2. Anonymous feedback through questionnaire at the end of the semester.

## COURSE POLICY

### MODE OF CORRESPONDENCE (email/ phone etc.)

E-mail: [shima@nitt.edu](mailto:shima@nitt.edu)/ Phone: +91-9447956884

### COMPENSATION ASSESSMENT POLICY

For those students who missed assessment 1 due to genuine reasons, compensation assessments will be conducted during III week of April 2020.

### ATTENDANCE POLICY (A uniform attendance policy as specified below shall be followed)

- **At least 75% attendance in each course is mandatory.**
- **A maximum of 10% shall be allowed under On Duty (OD) category.**
- Students with **less than 65% of attendance** shall be prevented from writing the final assessment and **shall be awarded 'V' grade.**

### ACADEMIC DISHONESTY & PLAGIARISM

- Possessing a mobile phone, carrying bits of paper, talking to other students, copying from others during an assessment will be treated as punishable dishonesty.
- Zero mark to be awarded for the offenders. For copying from another student, both students get the same penalty of zero mark.
- The departmental disciplinary committee including the course faculty member, PAC chairperson and the HoD, as members shall verify the facts of the malpractice and award the punishment if the student is found guilty. The report shall be submitted to the Academic office.

The above policy against academic dishonesty shall be applicable for all the programs.

## ADDITIONAL INFORMATION

The respective faculty will be available for consultation at times as per the intimation by the faculty.

## FOR APPROVAL

  
Course Faculty 04/02/2020

CC-Chairperson 

HOD 



**Guidelines**

- a) The number of assessments for any theory course shall range from 4 to 6.
- b) Every theory course shall have a final assessment on the entire syllabus with at least 30% weightage.
- c) One compensation assessment for absentees in assessments (other than final assessment) is mandatory. Only genuine cases of absence shall be considered.
- d) The passing minimum shall be as per the regulations.

B.Tech. Admitted in				P.G.
2018	2017	2016	2015	
35% or (Class average/2) whichever is greater.		(Peak/3) or (Class Average/2) whichever is lower		40%

- e) Attendance policy and the policy on academic dishonesty & plagiarism by students are uniform for all the courses.
- f) Absolute grading policy shall be incorporated if the number of students per course is less than 10.
- g) Necessary care shall be taken to ensure that the course plan is reasonable and is objective.