

DEPARTMENT OF CIVIL ENGINEERING NATIONAL INSTITUTE OF TECHNOLOGY TIRUCHIRAPPALLI - 620 015, TAMIL NADU, INDIA

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Name of the programme and specialization		RSE PLAN (PA ch / Civil Engine						
Course Title	Envi	ronmontal Frain						
Course Code			eering Laboratory					
Course Code	CEL	R13	No. of Credits	2				
Course Code of Pre-requisites Session	-	-						
	July	/ 2019	Section/s	A & B				
Name of the Faculty E-mail	Mr.Samjebadurai.C		Department	Civil Engineering				
		eba@nitt.edu	Telephone No.	9940839398				
Course Coordinator(s)(if, applicable)	-							
L-IIIdii	-		Telephone No.	-				
Course Type		Core						
		Elective						
COURSE CONTENT (Approved in Senate	V	√ Laboratory						

Physical characteristics of water - turbidity, suspended solids. Chemical characteristics of water - pH, hardness, alkalinity, chlorides, sulphates, iron, residual chlorine, total solids, dissolved solids, organic and inorganic solids, DO, BOD, COD. Optimum coagulant dose-Bacteriological tests - Microscopic tests.

COURSE LEARING OBJECTIVES

- 1. To analyse the physical and chemical characteristics of water and wastewater
- 2. To quantify the chemical requirement for turbidity removal
- 3. To familiarize the methods to estimate the organic strength of wastewater
- 4. To study the growth of microorganisms and its quantification

COURSE OUTCOMES (CO)

Course Outcomes (CO) & Aligned Programme Outcomes(PO)

After successful completion of the course, the students should be capable to

	the state of the s										
	CO1	Apply different analysis techniques for the measurement of physical and chemical parameters of									
		wastewater									
	CO2	Quantify the pollutant concentration in water and wastewater									
	CO3	Recommend the degree of treatment required for the water and wastewater									
		Assess the microbial contamination in water									
1											

CO/PO	1	2	3	Δ	5	6	7	0				
00/10	1	-	1	+ -	3	U	/	8	9	10	11	12
CO1	V	V	V	\ \			V					+
CO/PO CO1 CO2 CO3 CO4	V	V		Land C		V	V				1	-
CO3	5 75	1	1	V same	And the state of	1 3 1-					<u> </u>	-
CO4		V		Sept Server	THE PARTY	1	V				-	

COURSE PLAN (PART II)

COURSE OVERVIEW

The course provides methods to estimate, analyze various constituents present in water and asses its compatibility for further utilities. And it also provides various techniques to quantify the pollution potential of wastewater.

	SE TEACHING AND LEARNIN	GACI	IVITIES		Mod	e o	f Delivery	
SI.No.	Week		Top		Lecture	1	Laborato	
1.	5 th week of	Det	ermination of pH and	exercise				
	July 2019							
	(3 Contact Hours)				Lecture	1	Laborator	
2.	2 nd week of	Det	ermination of Chlorid	es	exercise		2000,010	
	August 2019				CACIOIO			
	(3 Contact Hours)		The second secon	And the second s	Lecture	1	Laborator	
3.	3 rd week of	Esti	mation of Total Hardr	ness	exercise	,	Laborator	
	August 2019				EXELCISO			
	(3 Contact Hours)				Lecture	1	Laborator	
4.	4th week of	Esti	mation of Calcium Ha	ardness	exercise	•	Laborator	
	August 2019				exercise			
	(3 Contact Hours)					-	Laborator	
5.	5 th week of	Det	ermination of Alkalinit	ty	Lecture	,	Laborator	
	August 2019				exercise			
	(3 Contact Hours)					-		
6.	1st week of	Dete	ermination of Sulphat	es	Lecture	1	Laborator	
0.	September 2019				exercise			
	(3 Contact Hours)							
7.	2 nd week of	Dete	ermination of Iron		Lecture	1	Laboratory	
•	September 2019				exercise			
	(3 Contact Hours)							
8.	3 rd week of	Dete	ermination of Total S	olids, Dissolved Solids,	Lecture	1	Laboratory	
0.	September 2019	Tota	Suspended Solid	exercise				
	(3 Contact Hours)	Soli		-,				
9.	4 th week of		ermination of Optimur	Lecture	1	Laboratory		
9.	September 2019	Den	similation of optimal	exercise				
	(3 Contact Hours)							
10.	1 st week of	Dete	armination of Dissolve	Lecture	1	Laboratory		
10.	October 2019	Dele	Determination of Dissolved Oxygen				,	
	(3 Contact Hours)				exercise			
44	2 nd week of	Dete	rmination of Riochem	Lecture	1	Laboratory		
11.	October 2019	Dele	etermination of Biochemical Oxygen Demand				,	
				exercise				
40	(Contact Hours)	Dete	ermination of Chemica	al Oxygen Demand	Lecture	1	Laboratory	
12.	3 rd week of October 2019	Dele	entination of offerno	exercise				
						CACIOISC		
	(3 Contact Hours)							
	4th - L - F		Δ	ssessment 3		-		
13.	4 th week of	1	Λ.	33633116111.0				
	October 2019	Ctor	dard Plate Count Tes	1	Lecture / F)em	onstration	
14.	5 th week of	Star	landard Flate Count Test			Lociaro y Domicrion and		
	October April 2019	1 1 1 4						
	(3 Contact Hours)	A 17 PM 14 PM		WEEKS WEEKSTEEL	STORY STORY			
	E ASSESSMENT METHODS	elginered a	Week / Date	Duration	% W	eiat	ntage	
SI.	Mode of Assessment	MARKET	Week / Date	Daration	,,			
No.			Weekly	NA		30%)	
1.	Assessment 1	- a a b	VVECKIY	10.	·			
	(Continuous Assessment on	each					1	
	practical session)		During the	NA	20%			
2.	Assessment 2		Session	2070				
	(Record Work)		4 th week of	45 Minutes	-	20%		
3.	Assessment 3		October, 2019	40 Milliards	•			
	(Test)		3 rd week of	30%				
4.	End Assessment		November, 2019	3070				
-7 .					assessed			

the attainment also)

It is proposed to take feedback from the students to evaluate the execution of the course

- Direct feedback from the students by face-to-face meeting individually and as the class as a whole.
- Feedback from the students during class committee meetings
- Exit survey from the students at the end of the session

The attendance will be taken in all the contact hours. Students are encouraged to attend all the classes without absence. Also, the students are encouraged to participate in various co-curricular and extracurricular activities to oprich the academic / campus life.

At least 75% attendance in the course is mandatory.

b) A maximum of 10% shall be allowed under On Duty (OD) category c) Students with less than 65% of attendance shall be prevented from writing the end assessment and shall be

COMPENSATION ASSESSMENT

a) Compensation Session will be provided on 2nd week of November, 2019 which is limited to only one experiment.

b) Students with valid reasons and prior permission are only allowed to appear for the compensation session.

ACADEMIC DISHONESTY AND PLAGIARISM

Academic Dishonesty

a) Possessing a mobile phone, carrying bits of paper, talking to other students, copying from others during an assessment will be treated as punishable dishonesty

b) Zero mark to be awarded for the offenders. For copying from another student, both students get the same

penalty of zero mark.

The department disciplinary committee constituted with the faculty member, PAC Chair person, and the HoD, as members shall verify the facts of the malpractice and award the punishment if the student found guilty,

Plagiarism

Turing someone else work as your own without proper consent

Providing incorrect information about the source of a quotation

ADDITIONAL COURSE INFORMATION

1. All the students are advised to check their NIT-T webmail regularly to know the updates.

2. Queries / Clarifications / Discussions (if required) may be E-mailed to me / contact me during working hours with prior intimation.

FOR APPROVAL

Mr. Samjebadurai.C Course Faculty

Dr. J. Karthikeyan Chairman (Class Committee) **HoD / Civil Engineering**