

DEPARTMENT OF CHEMISTRY

| | COURSE PLA | N – PART I | |
|---|-----------------------------------|-----------------|---|
| Name of the programme and specialization | M.Sc.(Chemistry) | | |
| Course Title | Lanthanide and Actinide Chemistry | | |
| Course Code | CH 634 | No. of Credits | 3 (Theory) |
| Course Code of Pre- requisite subject(s) | Nil | | |
| Session | July 2019 | | |
| Name of Faculty | Dr. A. Sreekanth | Department | Chemistry |
| Official Email | sreekanth@nitt.edu | Telephone No. | 9489551851 |
| Name of Course Coordinator | Dr. V.M.Biju | | |
| Official E-mail | vmbiju@nitt.edu | Telephone No. | +91-431-2503638(O) +91-9443843076(M) |
| Course Type | Core course | ZElective cours | se |

Syllabus (approved in BoS)

Lanthanides: Occurrence – Ores- Extraction and separation – The Lanthanide contraction – Electronic configuration – shapes of f – orbitals – ionization energies – simple binary compounds of lanthanides

Lanthanides: Coordination chemistry – Coordination numbers – stability and oxidation states - Magnetic Properties – Electronic Spectra – Luminescence Spectra – NMR Applications – and Imaging- EPR Spectroscopy

Organolanthanide Chemistry: Stability -+3 oxidation state - Alkyls and aryls - Cyclopentadienyls - hydrides- other oxidation states and their organometallic complexes - carbonyl compounds of Sc-Y & Pr

Actinides: Occurrence – Synthesis – Extraction and isolation – Characteristics of the actinides – reduction potentials – relativistic effects – binary compounds of actinides – coordination chemistry of actinides – stability – structure and coordination number

Actinides: Electronic and magnetic properties of actinides – spectra – Organoactinides – cylopentadienyls- carbonyls – synthesis of transactinides – naming.

COURSE OBJECTIVES

The student will be able to in detail understand the chemistry and reactivity of lanthanides and actinides, their properties and applications.



| MAPPING OF COs with POs | | |
|--|--|--|
| Course Outcomes | Programme Outcomes (PO) (Enter Numbers only) | |
| 1. Learn about Lanthanides and their extraction | 1,2,3, 4,5,6,8,10 | |
| 2. Understand Lanthinde Chemistry and spectroscopy | 1,2,3, 4,5,6,10 | |
| 3. Learn about actinides and their extraction | 1,2,3, 4,5,6,10 | |
| 4. Learn about actinides and their extraction | 1,2,3, 4,5,6,10 | |

COURSE PLAN - PART II COURSE OVERVIEW This 3 credit course is for theory. Three theory classes will be conducted per week. COURSE TEACHING AND LEARNING ACTIVITIES (Add more rows) S.No. Week Topic Mode of Delivery Lanthanides: Occurrence - Ores-C&T, PPT 1 1 Extraction and separation – The C&T, PPT Lanthanide contraction 2 Π - Electronic configuration - shapes of C&T, PPT f - orbitals - ionization energies -3 IIIsimple binary compounds of lanthanides Lanthanides: Coordination chemistry C&T, PPT - Coordination numbers -4 IV stability and oxidation states -C&T, PPT Magnetic Properties – Electronic 5 V Spectra -Luminescence Spectra -C&T, PPT 6 VI



| 7 | VII | NMR Applications – and Imaging- EPR Spectroscopy | C&T, PPT |
|----|------|---|----------|
| 8 | VIII | Organolanthanide Chemistry: Stability -+3 oxidation state – Alkyls and aryls | C&T, PPT |
| 9 | IX | Cyclopentadienyls – hydrides- other oxidation states and their organometallic complexes – carbonyl compounds of Sc-Y & Pr | C&T, PPT |
| 10 | X | Actinides: Occurrence – Synthesis – Extraction and isolation – | C&T, PPT |
| 11 | ΧI | Characteristics of the actinides – reduction potentials – relativistic effects – | C&T, PPT |
| 12 | XII | binary compounds of actinides – coordination chemistry of actinides – stability – structure and coordination number | C&T, PPT |
| 13 | XIII | Electronic and magnetic properties of actinides – spectra – | C&T, PPT |
| 14 | XIV | Organoactinides – cylopentadienyls- carbonyls – | C&T, PPT |
| 15 | XV | synthesis of transactinides – naming. | C&T, PPT |

COURSE ASSESSMENT METHODS (shall range from 4 to 6)

Theory

| S.No. | Mode of Assessment | Week/Date | Duration | % Weightage |
|-------|--------------------|-------------------------|------------|-------------|
| 1 | Assignment | I week of September | One week | 5 |
| 2 | Test I | II week of September | 60 minutes | 20 |
| 3 | Seminar | III week of October | One week | 5 |
| 4 | Test II | IV week of October | 60 minutes | 20 |



| СРА | Compensation Assessment | II week of November | 60 minutes | 20 |
|-----|-------------------------|------------------------|------------|----|
| 5 | Final Asessment | I week of December | 3 hours | 50 |

COURSE EXIT SURVEY

- 1. Feedback from students during class committee meetings.
- 2. Anonymous feedback through questionnaire at the end of the semester.

COURSE POLICY (including compensation assessment to be specified)

MODE OF CORRESPONDENCE (email/ phone etc): Through mobile phone COMPENSATION ASSESSMENT POLICY

- 1. This assessment is for those students who missed Test I or Test II due to genuine reasons
- 2. Compensation assessment will be conducted during the II week of November 2019.

ATTENDANCE POLICY (A uniform attendance policy as specified below shall be followed)

- At least 75% attendance in each course is mandatory.
- > A maximum of 10% shall be allowed under On Duty (OD) category.
- > Students with less than 65% of attendance shall be prevented from writing the final assessment and shall be awarded 'V' grade.

ACADEMIC DISHONESTY & PLAGIARISM

- Possessing a mobile phone, carrying bits of paper, talking to other students, copying from others during an assessment will be treated as punishable dishonesty.
- > Zero mark to be awarded for the offenders. For copying from another student, both students get the same penalty of zero mark.
- ➤ The departmental disciplinary committee including the course faculty member, PAC chairperson and the HoD, as members shall verify the facts of the malpractice and award the punishment if the student is found guilty. The report shall be submitted to the Academic office.
- > The above policy against academic dishonesty shall be applicable for all the programmes.

ADDITIONAL INFORMATION, IF ANY

The faculty will be available for consultation at times as per the intimation by the faculty.

| FOR APPROVAL | | | |
|----------------|-----------------|-------|-----|
| Course Faculty | CC- Chairperson | All . | HOD |
| | | | |

Page 4 of 6