DEPARTMENT OF CHEMISTRY

NATIONAL INSTITUTE OF TECHNOLOGY, TIRUCHIRAPPALLI

COURSE PLAN – PART I				
Name of the programme and specialization	M. Sc. Chemistry (I-Semester)			
Course Title	Nuclear Chemistry			
Course Code	CH 635 No. of Credits 3 (Theory)			
Course Code of Pre- requisite subject(s)				
Session	July 2018	Section (if, applicable)	A / B	
Name of Faculty	Dr. Seenuvasan Vedachalam	Department	Chemistry	
Email	vseenu@nitt.edu	Telephone No.	8220462456	
Name of Course Coordinator(s) (if, applicable)	Dr. Seenuvasan Vedachalam			
E-mail	vseenu@nitt.edu	Telephone No.	8220462456	
Course Type	Core course	Elective co	ourse	

Syllabus (approved in BoS)

Discovery: Types of decay - decay kinetics: decay constant, half-life period, mean life parent - daughter decay - growth relationships - secular and transient equilibrium - units of radioactivity - alpha, beta and gamma decay: Theory of decay, energies and properties -artificial radioactivity - detectors: Ionization chamber, electron pulse counters, scintillation detectors, semiconductor, detectors, thermo luminescence detectors and neutron detectors.

Types of nuclear reactions: Bethe notation, the compound nucleus theory - reaction cross – section - transmutation reactions, elastic and inelastic scattering, spallation, fragmentation, stripping and pick-up, fission, fusion, photonuclear reactions, thermonuclear reactions. The fission energy - reproduction factor - classification of reactors - based on moderators, coolent, phase of fuel and generation - principle of thermal nuclear reactors. The four-factor formula - reactor power - critical size of a thermal reactor - excess reactivity and control - breeder reactor - reprocessing of spent fuels - nuclear waste management - safety culture - active and passive safety, containment building, nuclear criticality safety, ionizing radiation protection - enforcement agencies.

Radiation chemistry: Passage of radiation through matter - units for measuring radiation absorption - radiation dosimetry - radiolysis of water - free radicals in water radiolysis - chemical dosimetry: Radiolysis of Fricke dosimeter solution - radiation - induced color centers in crystals - Effects of radiation with matter: Radiolysis of inorganic gases, organic gases, organic compounds, solids, and polymers - Annealing of radiation damage.

Application of radioisotopes: Probing by isotopes, reactions involved in the preparation of radioisotopes, the Szilard-Chalmer's reaction - radiochemical principles in the use of tracers - applications of radioisotopes as tracers - chemical investigations, analytical applications, agricultural and industrial applications - neutron activation analysis - carbon and rock dating - use of nuclear reactions - radioisotopes as source of electricity - nuclear medicines.

References:

1. W. Loveland, D. Morrissey, G. Seaborg. Modern Nuclear Chemistry, Wiley- Interscience, Hoboken, NJ, 2006.

2. Arnikar, H. J., Essentials of Nuclear Chemistry, 4th Ed., New Age International Publishers Ltd., New Delhi, 1995.

3. K. H. Lieser, Nuclear and Radiochemistry, 2nd revised ed., Wiley-VCH, Berlin, 2001.

4. G. Choppin, J. O Liljenzin and J. Rydberg. Radiochemistry and Nuclear Chemistry. 3rd ed. Butterworth-Heinemann, Oxford, 2002.

5. G Friedlander, GW Kennedy, ES Macias and JM Miller. Nuclear and Radiochemistry. 3rd ed., John Wiley & Sons, New York, 1981.

6. S. Glasstone, Source Book on Atomic Energy, Krieger Pub Co, 3rd Ed, 1979.

COURSE OBJECTIVES

COURSE OVERVIEW

To know the concept of Nuclear chemisty in mordern approach. Also students will be studying the various nuclear reaction, kinetics and application towards various industries such as agricultural, medicine and nuclear reactor for energy application.

COURSE OUTCOMES (CO)

Co	urse Outcomes	Aligned Programme Outcomes (PO)
1.	Students will become familiar with the basics of Nuclear chemistry	
2.	In addition, they will be learning the Types of decay, Theory of decay and know the knowledge of various radiation detector.	
3.	Students gain knowledge on various types of Nuclear reaction and how to handle safely from radiation hazard with the help of various physical parameters.	
4.	Students will become familiar with how the nuclear radiation is useful to the human societies	
5.	This course is highly useful because students can get better knowledge in applied sciences.	

COURSE PLAN – PART II

This is an elective course offered for the M. Sc. Students. Three credits are awarded for the course. Three lectures will be conducted every week by a faculty member in Chemistry dept.

COURSE TEACHING AND LEARNING ACTIVITIES

S.No.	Week/Contact Hours	Торіс	Mode of Delivery
1	l-Week Aug/2018	Discovery: Types of decay -decay kinetics:	Chalk and Talk
2	ll-Week Aug/2018	decay constant, half-life period, mean life parent - daughter decay - growth relationships - secular and transient equilibrium	Chalk and Talk

3	III-Week Aug/2018	units of radioactivity - alpha, beta and gamma decay: Theory of decay, energies and properties -artificial radioactivity -	Chalk and Talk
4	IV-Week Aug/2018	detectors: Ionization chamber, electron pulse counters, scintillation detectors, semiconductor, detectors, thermo luminescence detectors and neutron detectors	Chalk and Talk
5	I-Week Sep/2018	Types of nuclear reactions: Bethe notation, the compound nucleus theory - reaction cross –section - transmutation reactions	Chalk and Talk
6	ll-Week Sep/2018	elastic and inelastic scattering, spallation, fragmentation, stripping and pick-up	Chalk and Talk & PPT
7	III-Week Sep/2018	fission, fusion, photonuclear reactions, thermonuclear reactions the fission energy - reproduction factor - classification of reactors - based on moderators, coolent, phase of fuel and generation	Chalk and Talk & PPT
8	IV-Week Sep/2018	- principle of thermal nuclear reactors. The four-factor formula - reactor power - critical size of a thermal reactor - excess reactivity and control - breeder reactor - reprocessing of spent fuels	Chalk and Talk & PPT
9	I-Week Oct/2018	nuclear waste management - safety culture - active and passive safety, containment building, nuclear criticality safety, ionizing radiation protection - enforcement agencies	Chalk and Talk & PPT
10	ll-Week Oct/2018	Radiation chemistry: Passage of radiation through matter - units for measuring radiation absorption - radiation dosimetry - radiolysis of water - free radicals in water radiolysis - chemical dosimetry:	Chalk and Talk & PPT
11	III-Week Oct/2018	Radiolysis of Fricke dosimeter solution – radiation – induced color centers in crystals – Effects of radiation with matter: Radiolysis of inorganic gases, organic gases, organic compounds, solids, and polymers – Annealing of radiation damage	Chalk and Talk & PPT
12	IV-Week Oct/2018	Application of radioisotopes: Probing by isotopes, reactions involved in the preparation of radioisotopes,	Chalk and Talk & PPT

13	l-Week Nov/2018	Szilard-Chalmer's reaction - radiochemical principles in the use of tracers - applications of radioisotopes as tracers - chemical investigations,	Chalk and Talk & PPT	
14 III-Week Nov/2018		analytical applications, agricultural and industrial applications - neutron activation analysis - carbon and rock dating - use of nuclear reactions - radioisotopes as source of electricity - nuclear medicine	Chalk and Talk & PPT	

COURSE ASSESSMENT METHODS (shall range from 4 to 6)

S.No.	Mode of Assessment	Week/Date	Duration	% Weightage
1	Assignment I	ll-Week Sep/2018	One week	5
2	Test I	II-Week Oct/2018	1hour	20
3	Assignment II	IV-Week Oct/2018	One week	5
4	Test II	II-Week Nov/2018	1hour	20
СРА	Compensation Assessment* Re-Test	IV Nov Dec/2018	1 hour	20
5	Final Assessment *	I-Week Dec/2018	3 hours	50

COURSE EXIT SURVEY (mention the ways in which the feedback about the course shall be assessed)

- 1. Feedback from students during class committee meetings.
- 2. Anonymous feedback through questionnaire at the end of semester.

COURSE POLICY (preferred mode of correspondence with students, compensation assessment policy to be specified)

MODE OF CORRESPONDENCE (email/ phone etc)

Phone;8220462456

<u>ATTENDANCE POLICY</u> (A uniform attendance policy as specified below shall be followed)

> At least 75% attendance in each course is mandatory.

- > A maximum of 10% shall be allowed under On Duty (OD) category.
- Students with less than 65% of attendance shall be prevented from writing the final assessment and shall be awarded 'V' grade.

THEORY:

Each student should score minimum of **40% is higher** to pass in this course **COMPENSATION ASSESSMENT POLICY**

Those who are absent for the assessment test on genuine grounds shall be given an opportunity only once for the retest with prior permission of the concern faculty member and head of department. The retest will be conducted before the end semester exam and portion will be all the five units.

ACADEMIC DISHONESTY & PLAGIARISM

> Possessing a mobile phone, carrying bits of paper, talking to other students, copying

from others during an assessment will be treated as punishable dishonesty.

- > Zero mark to be awarded for the offenders. For copying from another student, both students get the same penalty of zero mark.
- The departmental disciplinary committee including the course faculty member, PAC chairperson and the HoD, as members shall verify the facts of the malpractice and award the punishment if the student is found guilty. The report shall be submitted to the Academic office.
- > The above policy against academic dishonesty shall be applicable for all the programmes.

ADDITIONAL INFORMATION

The respective faculty will be available for consultation at times as per the intimation by the faculty.

FOR APPROVAL

Course Faculty N. Sconwelco, CC-Chairperson_____ HOD___ intur

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