



# National Institute of Technology, Tiruchirappalli–620015

## Department of Chemistry

<b>COURSE PLAN</b>			
<b>Course Title</b>	Coordination Chemistry and its Applications		
<b>Course Code</b>	CHMI10	<b>No. of Credits</b>	3
<b>Department</b>	Chemistry	<b>Faculty</b>	Dr. G. Venkatesa Prabhu Dr. Shima P. Damodaran Dr. S. Amudhan Senthana
<b>Pre-requisites Course Code</b>	NA		
<b>Course Coordinator (if, applicable)</b>	Dr. G. Venkatesa Prabhu		
<b>E-mail</b>	venkates@nitt.edu shimapd@gmail.com amudhansenthana.s@gmail.com	<b>Mobile No.</b>	+91 9894080616 +91 9447956884 +91 9840699833
<b>Course Type</b>	Minor/Elective course		
<b>COURSE OVERVIEW</b>			
This is a minor three credit course offered to B.Tech. students. Three theory classes will be conducted per week.			
<b>COURSE OBJECTIVES</b>			
Enabling the students to grasp the basics and advances in the area of coordination chemistry and to gain a clear picture on the principles and applications which underlie the subject.			
<b>COURSE OUTCOMES (CO)</b>			
The students will familiarize themselves in <ul style="list-style-type: none"><li>✓ Learning about the basic concepts and theories of bonding in coordination chemistry.</li><li>✓ Understanding the spectroscopic and magnetic characteristics of metal complexes.</li><li>✓ Learning about the stability factors and reaction mechanisms of metal complexes.</li><li>✓ Having a thorough picture of the various biological important coordination compounds.</li><li>✓ Having an overview of the applications of coordination chemistry in biological and industrial processes.</li></ul>			

<b>COURSE TEACHING AND LEARNING ACTIVITIES</b>			
<b>S.No</b>	<b>Week</b>	<b>Topic</b>	<b>Mode of Delivery</b>
1	I week of Jan 2018	<b>UNIT-I</b> <b>Basic concepts and theories of coordination compounds:</b> Introduction - Nomenclature of coordination compounds - Electronic theory of complex compounds - valence bond theory - Crystal field theory - assumptions of crystal field theory	C&T, PPT
2	II week of Jan 2018	Crystal field splitting in octahedral and tetrahedral geometries - CFSE, factors affecting the magnitude of $10 Dq$ , Jahn-Teller distortion - Merits and limitations of CFT	C&T, PPT
3	III week of Jan 2018	Molecular orbital theory of selected octahedral and tetrahedral complexes. Polynuclear complexes - non-ionic complexes - coordination polymers - isomerism.	C&T, PPT
4	IV week of Jan 2018	<b>UNIT-II</b> <b>Spectral and magnetic characteristics of metal complexes:</b> Absorption spectra of complexes - interpretation, term symbols and splitting of terms in free atoms, selection rules for electronic transitions	C&T, PPT
5	I week of Feb 2018	Orgel diagram, electronic spectra of $d^x$ complexes, Charge transfer spectra, CT versus d-d transitions	C&T, PPT
6	II week of Feb 2018	Magnetic properties of metal complexes - determination of magnetic susceptibilities - magnetic criterion of bond type - orbital contribution to magnetic moment.	C&T, PPT
7	III week of Feb 2018	<b>UNIT-III</b> <b>Stability and reactions of metal complexes:</b> Thermodynamic stability and Kinetic stability of complexes - stability constants - Factors affecting the stability of coordination compounds	C&T, PPT
8	IV week of Feb 2018	Determination of stability constants. Labile and inert coordination compounds - Ligand substitution reactions ( $SN_1$ , $SN_2$ and $SN_1CB$ ) - anation reaction	C&T, PPT
9	I week of Mar 2018	Electron transfer reaction - OSM - ISM. Substitution reactions in square planar complexes - trans effect - theories and applications of trans effect.	C&T, PPT
10	II week of Mar 2018	<b>UNIT-IV</b> <b>Biological coordination compounds:</b> Transport and storage of oxygen - hemoglobin and myoglobin- hemeythrin and hemocyanine - syntheticoxygen carriers	C&T, PPT
11	III week of Mar 2018	Electron transfer - Cytochromes, Rubredoxins and Ferredoxins. Catalysis - Catalase - blue copper proteins	C&T, PPT

12	IV week of Mar 2018	Metalloenzymes. Photosynthesis - chlorophyll - vitamin B <sub>12</sub> and its biological functions.	C&T, PPT
13	I week of Apr 2018	<b>UNIT-V</b> <b>Applications of coordination chemistry:</b> Analytical chemistry: Inorganic qualitative analysis, complexometric titrations, complexes in colorimetry, gravimetry and separation of metals. Medicinal chemistry: complexation in food poisoning and metal complexes in therapy	C&T, PPT
14	II week of Apr 2018	Anticancer drugs, anti-arthritis drug of gold drugs and radiodiagnostic agents. Electrochemistry: Electro deposition of specific metals, Chromium, Copper, Nickel, Precious metal.	C&T, PPT
15	III week of Apr 2018	Dyes and pigments: Metal complexes of azo compounds - azomethines. Industrial processes: catalytic activation - hydrogenation, hydroformylation and oxidation of olefins.	C&T, PPT

#### COURSE ASSESSMENT METHODS

S.No.	Mode of Assessment	Week/Date	Duration	% Weightage
1	Assignment/Quiz/Seminar	I week of February	Depends on the activity	5
2	Cycle Test - I	III week of February	1 h	20
3	Assignment/Quiz/Seminar	III week of March	Depends on the activity	5
4	Cycle Test - II	I week of April	1 h	20
5	End Semester	IV week of April	3 h	50

#### ESSENTIAL READINGS : Textbooks, reference books Website addresses, journals, etc

1. J. D. Lee, *Concise Inorganic Chemistry*, 5<sup>th</sup> Edition, Chapman and Hall, London, 1996.
2. G. O. Spessard and G. L. Miessler, *Organometallic Chemistry*, 2<sup>nd</sup> Edition, Oxford University Press.
3. F. A. Cotton, G. Wilkinson, C. A. Murillo and M. Bochmann; *Advanced Inorganic Chemistry*, 6<sup>th</sup> Edition. Wiley, 1999.
4. J. E. Huheey, E. A. Keiter and R. L. Keiter, *Inorganic Chemistry, Principles of Structure and Reactivity*, 4<sup>th</sup> Edition, Harper Collin College Publishers, 1993.
5. *Comprehensive Coordination Chemistry. The Synthesis, Reactions, Properties and Application of Coordination Compounds*, Volume 6, Sir Geoffrey Wilkinson, Robert D. Gillard, Jon A McCleverty, 1987 Pergamon Books Ltd. England.
6. S. J. Lippard and J. M. Berg, *Principles of Bioinorganic Chemistry*, Univ. Science Books, 1994.

**COURSE EXIT SURVEY (mention the ways in which the feedback about the course is assessed and indicate the attainment also)**

1. Feedback from students during class committee meetings.
2. Anonymous feedback through electronic media.

**COURSE POLICY (including plagiarism, academic honesty, attendance, etc.)**

1. Tests I and II will be conducted in regular class.
2. Plagiarism is forbidden and will be dealt sternly.
3. 75% attendance is mandatory for writing the end semester examination.
4. A retest will be conducted for those who missed the CT - I or II or both for genuine reasons during the III week of April 2018, covering the syllabus of Tests I and II.

**ADDITIONAL COURSE INFORMATION**

The course faculty will be available for consultation at times as per the intimation by the faculty.

**FOR SENATE'S CONSIDERATION**

**Course Faculty**

**CC-Chairperson**

**HOD**