Department of Chemistry

NATIONAL INSTITUTE OF TECHNOLOGY, TIRUCHIRAPPALLI

CSE-B

		COURSE PLAN	(For CSE B)			
Course	Title	Chemistry II		ACCOUNTS OF A PARTITION		
Course	Code	CHIR 13	No. of Credits	4 (7	4 (Theory-3 + Lab-1)	
Departn	nent	Chemistry	Faculty	Dr.	Dr. L. Cindrella	
Progran	nme	B.Tech. (CSE B)				
Pre-requ		NIL				
Course						
	Coordinator(s)	Dr. L.Cindrella (Theory)				
(if, appli E-mail	cable)	Dr.S.Velmathi (Lab) cind@nitt.edu	Telephone No. 2503634			
L-IIIaii		velmathis@nitt.edu	Telephone No. 2503634 2503640			
Course	Туре	Core course Elective course				
COURS	E OVERVIEW					
theory (3	3 credit) and prac		neory classes w		ourse is a combination of conducted per week and	
COURS	E OBJECTIVE					
		principles of electroche ech. (CSE B) students.	emistry, corrosi	on, Ba	tteries, solid state and	
COURS	E OUTCOMES (C	0)				
✓ ir ✓ c ✓ p	rauses, consequer principles, operation preparation of sem	amiliar with the trochemistry and its appli nces and protection meth on and uses of batteries iconductors and its purifi plications of polymers and	ods of corrosion	n		
COURS	E TEACHING AN	D LEARNING ACTIVITII	ES			
S.No.	Week		Topic		Mode of Delivery	

S.No.	Week	Topic	Mode of Delivery
1	III week of Jan	Unit-I Conductivity of electrolytes, specific, molar and equivalent conductivity, Nernst equation, EMF series	C&T, PPT
2	IV week of Jan	Hydrogen, calomel & glass electrodes, electrolytic & Galvanic cells, cell EMF (measurement & applications)	C&T, PPT
3	I week of Feb	Weston standard cell, reversible and irreversible cells, concentration cell, electrode and	C&T, PPT FC, TPS

		electrolyte concent	tration cell.	
4	II week of Feb	Unit-II Dry & wet corrosion, mechanisms, types of corrosion (DMC, DAC, stress, intergranular, atmospheric and soil)		C&T, PPT
5	III week of Feb	significance, factor protection from cor		C&T, PPT
6	IV week of Feb		roplating, electrolessplating, cathodic ction, chemical conversion and organic	
7	I week of March	Unit-III Different types of batteries-Primary, secondary & Flow battery. Working principle and uses of Laclanche cell, Alkaline, nicad, lithium and mercury battery		C&T, PPT
8	II week of March	Theory, working ar Different types of f oxygen, PEFC and	C&T, PPT	
9	III week of March	Charging & discharging principle, operation and uses of Lead acid storage cell. Working principle of solar battery		FC, TPSC&T, PPT
10	IV week of March	Unit-IV Types of solids, Close packing of atoms and ions. BCC, FCC structures of rock salt & CsCl		C&T, PPT
11	I week of April	Introduction about spinels-normal and Stoichiometric defe chalcogen semicor	C&T, PPT	
12	II week of April	Non-elemental semiconducting materials, Steps involved in the preparation of semiconductors. Semiconductor devices- p-n junction diodes.		C&T, PPT FC, TPS
13	III week of April	Unit-V Polymers, nomenclature, tacticity, polymerization processes, mechanisms, types of polymerization, classification		C&T, PPT
14	IV week of April	Effect of structure on properties, moulding, important polymers, synthesis & properties		C&T, PPT
15	I week of May	Molecular mass determination, Synthesis & applications of some commercially important polymers, conducting polymers		C&T, PPT FC, TPS
	SE ASSESSMENT ME		Duration	0/ \\\a:b4=
S.No.	Mode of Assessment	Week/Date	Duration	% Weightage
Theory				
1	TPS (Quiz)	I week of Feb	15 minutes	5
2	Test I	20.02.2017	60 minutes	15
3	TPS (Quiz)	IV week of Feb	15 minutes	5
4	TPS (Quiz)	III week of March	15 minutes	5
				15
5	Test II	03.04.2017	60 minutes	15

7	End semester	II week of May	3 hours	50	
Practical					
8	Regular class	All practical	3 hours per experiment	25	
	experiments	classes			
Theory (75) + Drectical (25) - Total (100)					

Theory (75) + Practical (25) = Total (100)

ESSENTIAL READINGS: Textbooks, reference books Website addresses, journals, etc

- 1. Engineering Chemistry, P.C. Jain & M. Jain, Dhanpat Rai Publishing Company, New Delhi, 2012
- 2. Principles of Physical Chemistry, B.R. Puri, L.R. Sharma & M.S. Pathania, Vishal Publishing Company, 2008
- 3. Textbook of Polymer Science, F.W. Billmayer, Wiley, New York, 1991
- 4. Concise Inorganic Chemistry, J.D. Lee, Chapman and Hall, 5th Edn., 1996.
- 5. A Textbook of Engineering Chemistry, S.S. Darer & S.S. Umare, S. Chand Publishing, 2011

COURSE EXIT SURVEY (mention the ways in which the feedback about the course is assessed and indicate the attainment also)

- 1. Feedback from students during class committee meetings.
- 2. Anonymous feedback through questionnaire at the end of the semester.

COURSE POLICY (including plagiarism, academic honesty, attendance, etc.)

- 1. 75% attendance (as on 5th May 2017) is compulsory for both Theory and Lab components.
- 2. Lab:
 - A. Each experiment will be evaluated for 5 marks.
 - B. There will be no final assessment for practical.
- C. One extra class will be conducted for those who missed any experiment due to ill health or OD reasons.
- 3. Theory:
- A. Those who have <75% attendance have to attend evening hour classes during the I week of May 2016 to become eligible to appear for assessment 7 (End semester).
- B. For those who missed Test I or Test II due to genuine reasons, retest will be conducted during the I week of May 2017.
- C. Attendance in any one the tests is compulsory for appearing in the final assessment (End Semester)

ADDITIONAL COURSE INFORMATION

The faculty will be available for consultation at times as per the intimation by the faculty.